

#Jenny



Finally I get this ebook, thanks for all these I can get now!

#Rio



Cool! I'am really happy

#Markus Jensen



I did not think that this would work, my best friend showed me this website, and it does! I get my most wanted eBook

#Hun Tsu



wtf this great ebook for free?!

#Che Salsa








My friends are so mad that they do not know how I have all the high quality ebook which they do not!

#Diego Butler



so many fake sites. this is the first one which worked! Many thanks

- PART A
- Which of the following compounds has pyramidal geometry?  
(A)  $\text{BCl}_3$  (B)  $(\text{CH}_3)_3\text{C}^+$  (C)  $(\text{CH}_3)_3\text{N}$  (D)  $\text{BF}_3$
  - Which of the following compounds exhibits optical isomerism?  
(A) 1-Chloropentane (B) 1,2-Dichloropentane  
(C) 3-Chloropentane (D) 2-Chloropentane
  - Identify the atom-economy reaction from the following:  
(A) Grignard reaction (B) Wittig reaction  
(C) Diels-Alder reaction (D) Friedel-Crafts reaction
  - Chiral molecule B has  $[\alpha]_D^{25} = +24$  for 40% optical purity with (R)-configuration. What will be  $[\alpha]_D^{25}$  of the molecule B with (S)-configuration having 100% optical purity?  
(A) -36 (B) +30 (C) -60 (D) -60
  - The correct order of stability of carbocations is:  
(A)  $\text{CH}_3^+ > (\text{CH}_3)_2\text{CH}^+ > \text{CH}_3\text{CH}_2^+ > (\text{CH}_3)_3\text{C}^+$   
(B)  $(\text{CH}_3)_3\text{C}^+ > (\text{CH}_3)_2\text{CH}^+ > \text{CH}_3\text{CH}_2^+ > \text{CH}_3^+$   
(C)  $(\text{CH}_3)_3\text{C}^+ > (\text{CH}_3)_2\text{CH}^+ > \text{CH}_3^+ > \text{CH}_3\text{CH}_2^+$   
(D)  $\text{CH}_3\text{CH}_2^+ > \text{CH}_3^+ > (\text{CH}_3)_2\text{C}^+ > (\text{CH}_3)_3\text{C}^+$
  - The IUPAC name of  is:  
(A) Vinylalcohol (B) 3-Ethylbut-3-en-1-ol  
(C) 2-Ethylbut-3-en-1-ol (D) 2-Ethylbutanol
  - Which of the following is aromatic?  
(A)  (B)  (C)  (D) 
  - The coordination number of a metal ion situated at the center of a square antiprism of ligand atoms is:  
(A) 2 (B) 4 (C) 6 (D) 8
  - The strongest base among the following is:  
(A)  $\text{AsH}_3$  (B)  $\text{PH}_3$  (C)  $\text{NH}_3$  (D)  $\text{SbH}_3$

2

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